Name of the Programme: M.Sc. Part-I (Biochemistry)

Course Code: CHB-501

Title of the Course: Analytical Biochemistry-I

Number of Credits: 4

Effective from AY: 2022-23

Pre-requisites	Students should have graduate level knowledge either in chemical	or life sciences
for the Course:	or should have qualified change of discipline test.	
Course	1. To introduce various bioanalytical techniques for separation an	d purification
Objectives:	of biomolecules.	
	 To develop concepts in techniques used for routine biochemica chromatography, spectrophotometry, centrifugation, electrophoresis. To evaluate the utility of various analytical techniques as a q quantitative tool. 	l work such as microscopy, ualitative and
Content:		No of hours
	1. General principles of analytical biochemistry	4
	a. Selection of valid methods for analysis, Instrumental	
	methods, physiological	
	methods, assessment of analytical method <mark>s.</mark>	
	b. Quality assurance in analytical biochemistry: quality control	
	and quality assessment,	
	c. Accreditation of laboratories: standard operating procedure	
	and good laboratory practice, sampling for analysis,	
	calibration and graphical representation of data.	
	2. Acid, bases and buffers	10
	 a. Units used in quantitative biochemical measurements: molarity, normality, parts per million and percentage by weight/ volume, concept of pH using pH electrode and other ion selective electrodes., Eh, acid-base associations. b. Buffers, buffering capacity, measurement of pH, mechanism of dissociation of macromolecules, dissociation constants, pKa, pl, solvents (eluotropic series), peroxide values, solubility and affinity constants. 	
	3. Colligative Properties	4
	 a. Definitions, Factors affecting and Physiological Applications of Osmosis. b. Measurement of osmotic pressure, Osmoregulation, Adsorption, Colloids, Surface Tension and Viscosity. c. Numerical Problems based on above concepts. 	
	4. Centrifugation:	8
	a. Principle of centrifugation, concepts of RCF, different types	
	of instruments and rotors.	
	b. Preparative, differential and density gradient	
	centrifugation, analytical ultra-centrifugation.	
	c. Determination of molecular weights and other	
	E Electrophoratic techniques:	10
	s. Lieurophoreuc techniques:	10

	а.	Principles of electrophoretic separation, Types of	
		electrophoresis including paper, cellulose, acetate/nitrate	
		and gel (introduction to concepts of slab gel, tube,	
		continuous and discontinuous, etc).	
	b.	Gel electrophoresis - types of gel, Agarose GE,	
		Polyacrylamide gel electrophoresis PAGE, SDS- PAGE,	
		Isoelectric Focusing and ampholytes, 2-D, native, gradient	
		gels, PFGE, DGGE, TGGE.	
	с.	Capillary electrophoresis - instrumentation, sample	
		introduction in CE, types of CE, electrophoretic mobility and	
		electroosmotic mobility, total mobility, efficiency and	
		resolution in CE column.	
	d.	Separation of neutral molecules by MEKC.	
	e.	Staining strategies and procedures: Coomassie Brilliant blue	
		R/G 250, Silver, Fluorescent stains Flamingo, Oriole, SYPRO-	
		Ruby; Stain-free gels.	
	f.	Examples of separation of biomolecules by electrophoresis.	
	6. So	lvent extraction	5
	a.	Basic principle, types of extractions and application.	
	b.	Separations based on a partitioning between phases based	
		on chemical nature and polarity of analyte.	
	с.	Introduction to Soxhlet apparatus, solid phase extraction,	
		microwave assisted extraction, ultrasound assisted	
		extraction, counter current extraction.	
	7. Di	alysis	5
	a.	Principles and applications of equilibrium dialysis and	
	a.	Principles and applications of equilibrium dialysis and ultrafiltration.	
	a. b.	Principles and applications of equilibrium dialysis and ultrafiltration. Dialysis and Concentration, reverse dialysis.	
	a. b. c.	Principles and applications of equilibrium dialysis and ultrafiltration. Dialysis and Concentration, reverse dialysis. Artificial membranes, semi-permeable membranes, Donnan	
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	a. b. c. d.	Principles and applications of equilibrium dialysis and ultrafiltration. Dialysis and Concentration, reverse dialysis. Artificial membranes, semi-permeable membranes, Donnan membrane equilibrium. Biological significance of osmosis and micelles.	14
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	ICT mode should be preferred. Sessions should be interactive in nature to	
	enable peer group learning.	
References/	1. K. Wilson, J. Walker, Principles and Techniques of Practical Biochemistry;	
Readings:	Cambridge University Press, 7 th Edition, 2010.	
	 G. D. Christian, P. K .Dasgupta, K. A. Schug, Analytical Chemistry, John Wiley & Sons, 7th Edition, 2013. 	
	3. M. V. Parakhia, R. S. Tomar, S. Patel, B. A. Golakiya, Molecular Biology and Biotechnology, Microbial Methods, New India, 2010.	
	 D. J. Homes, H. Peck, Analytical Biochemistry, Pearson education Limited. 1998. 	
	 A. Skoog Douglas, F. James Holler, Stanley R. Crouch, Principles of Instrumental Analysis. 7th Edition. Cengage Learning, 2016. 	
	6. D. J. Holme., H. Peck, Analytical Biochemistry, 3 rd Edition, Prentice Hall 1998.	
Course	1. Students will be able to explain the principles of various separation	
Outcomes:	techniques	
	2. Students will be in a position to differentiate between various analytical	
	techniques for separation and purification of biomolecules based on their principles	
	 Students will be able to choose appropriate separation technique and isolate and purify biomolecules. 	
	 Students will be able to apply the knowledge of these techniques for designing various experiments in research and development 	