

Programme: M. Sc. (Marine Sciences)

Course Code: MSC 462 **Title of the Course:** Estuarine Chemistry

Number of Credits: 01

Effective from AY: June 2018-19

Prerequisites for the course:	Should have undergone the course Marine Chemistry (MSC 162).	
Objective:	This course develops concepts about the chemistry of the estuarine environment that concerns the study of the properties and interactions of the substances present in the estuarine environment.	
Content:	Salinity distribution in estuaries – a chemical perspective, flushing time, mixing and diffusion dispersal of pollutants in estuaries and near shore areas – Conservative and non – conservative properties of dissolved constituents during estuarine mixing – Behaviour of dissolved oxygen, pH and major elements in estuarine water.	12 hours

Pedagogy:	Lectures/tutorials/assignments/self-study.	
References/ Readings	1. Estuarine Chemistry, 1976 - Burton, J.D. and Liss, P.S., Academic Press. 2. Practical Estuarine Chemistry, 1985 – Head, P.C., Cambridge University Press. 3. Chemistry and Biogeochemistry of Estuaries, 1980 – Olausson, E. and Cato, I., John Wiley & Sons. 4. Chemical Oceanography (Vol.7), 1978, Riley, J.P. and Chester, R., Academic Press. 5. Waves, Tides and Shallow-Water Processes, 1991, 2005 – The Open University. 6. Coastal and Estuarine Sediment Dynamics, 1986 – Dyer, K.R., Wiley. 7. Estuarine Hydrography and Sedimentation, 1980 – Dyer, K.R., Cambridge University Press. 8. Biogeochemistry of Marine Dissolved Organic Matter, 2002–D.A.Hansell and C. A. Carlson., Academic Press. 9. Biogeochemistry of Estuaries, 2007 – Thomas S. Bianchi, Oxford University Press. 10. Treatise on Estuarine and Coastal Science - Vol. 4: Geochemistry of Estuaries and Coasts, Vol. 5: Biogeochemistry, 2012, E. Wolanski and D. McClusky, Elsevier Inc.	
Learning Outcomes	1. Provide a comprehensive understanding of the properties and interactions of the substances present in the estuarine environment. 2. Explain the key processes operating in the estuarine environment. 3. Explain the importance of dissolved O₂, pH and the CO₂ problem. 4. Explain the biogeochemical cycling of major seawater constituents from the perspective of the global biogeochemical cycling of elements.	