

**Programme: M. Sc. Part-I (Chemistry)**

**Course Code: ICO-401**

**Title of the Course: Topics in Inorganic Chemistry & Environmental Chemistry**

**Number of Credits: 03**

**Effective from AY: 2018-19**

<b>Prerequisites for the course:</b>	Student should have studied the courses in chemistry at F.Y. B.Sc., S.Y.B.Sc. and T.Y.B.Sc. levels and / or CHIC-401 course so as to have basic knowledge of Inorganic / environmental chemistry.	No. of lectures
<b>Course Objectives:</b>	<ol style="list-style-type: none"><li>1. To provide fundamental aspects of transition &amp; inner transition metals &amp; their compounds.</li><li>2. To provide knowledge of main group elements of the periodic table &amp; their compounds</li><li>3. To introduce various global phenomenon's of atmosphere &amp; environment, follow directive of the Supreme Court in 1993 to introduced environmental education at all levels, have a fair knowledge on the various global activities to justify permissible or adverse, so that future generation are not adversely affected.</li></ol>	
<b>Course Outcomes:</b>	<ol style="list-style-type: none"><li>1. Students should be in position to understand fundamentals / usefulness of transition &amp; inner transition metals.</li><li>2. Students should be in position to understand chemistry main group elements.</li><li>3. Students shall be aware of the maintenance of healthy living atmosphere on the globe.</li></ol>	
<b>Content:</b>	<p style="text-align: center;"><b>SECTION-I</b></p> <p><b>1. Chemistry of transition &amp; inner transition elements</b></p> <p>1.1 Transition elements: IUPAC definition of transition elements, occurrence, physical &amp; chemical properties, noble character, metal oxides &amp; oxido complexes, examples of metal-metal bonded clusters.</p> <p>1.2 Inner transition elements: Lanthanides, occurrence, properties, oxidation states, electronic structure, colour and spectra, magnetic properties, lanthanide contraction, compounds of lanthanides. Actinoid chemistry, general trends.</p> <p><b>2. Main group elements and their compounds</b></p> <p>2.1 Boron group: Compounds of boron:- borazine and boron nitride, synthesis, properties, structure &amp; bonding. Borates: classification, structures &amp; examples.</p> <p>2.2 Carbon group: Allotropes of carbon including C<sub>60</sub>, intercalation compounds of graphite, carbides. Compounds of silicon: silicates, zeolites &amp; silicones.</p> <p>2.3 Nitrogen group:- Introduction: oxides &amp; oxyacids of nitrogen. 2.4 Oxygen group: oxyacids &amp; oxohalides of S, S<sub>4</sub>N<sub>4</sub> ring compounds: synthesis, properties, structure &amp; bonding.</p>	<p>9 hr</p> <p>9 hr</p>

	<p style="text-align: center;">SECTION-II</p> <p><b>1. Atmosphere</b> Structure and properties of the atmosphere, composition of atmosphere and vertical temperature behaviour, lapse rate and temperature inversion.</p> <p><b>2. Air Pollution</b> Classification of air pollutants and photochemical reactions in the atmosphere Common air pollutants (e.g. CO, NO<sub>x</sub>, SO<sub>2</sub>, hydrocarbons and particulates) (a) sources (b) physiological and environmental effect (c) monitoring, d) various remedial &amp; technological measures to curb pollution. Air quality standards.</p> <p><b>3. Water pollution</b> Importance of buffer &amp; buffer index in waste water treatments. Chemical, physical &amp; biological characteristics of water pollution, specific &amp; non-specific characterization of water. DO, BOD, COD, and chlorine demand, typical water treatment &amp; waste water treatment (Municipal).</p> <p><b>4. Treatment of Industrial wastes</b> Electroplating industry, fertilizer industry and pharmaceuticals industries.</p> <p><b>5. Biogeochemical cycles:</b> Carbon and Nitrogen cycles nature</p>	<p>2 hr</p> <p>7 hr</p> <p>5 hr</p> <p>2 hr</p> <p>2 hr</p>
<b>Pedagogy:</b>	Mainly lectures / tutorials. Seminars / assignments / presentations / self-study or a combination of some of these could also be used to some extent.	
<b>Text books / reference books</b>	<p>1. P.W. Atkins, T. Overton, J. Rourke, M. Weller, &amp; F. Armstrong, <i>Shriver &amp; Atkins Inorganic Chemistry</i>, Oxford publications, 2009, 5<sup>th</sup> Ed.</p> <p>2. J. E. Huheey, E. A. Keiter, R. L. Keiter &amp; O. K. Medhi, <i>Inorganic Chemistry: Principles of Structure &amp; Reactivity</i>, Pearson, 2011, 4<sup>th</sup> Ed.</p> <p>3. F. A. Cotton, G. Wilkinson &amp; P. L. Gaus, <i>Basic Inorganic Chemistry</i>, Wiley, 2008 (reprint), 3<sup>rd</sup> Ed.</p> <p>4. N.N. Greenwood and A. Earnshaw, <i>Chemistry of the Elements</i>, Pergamon Press, Exetr, Great Britain. 1984.</p> <p>5. J.D. Lee, <i>Concise Inorganic Chemistry</i>, Wiley, 2008, 5<sup>th</sup> Ed.</p> <p>6. A.V. Salker, <i>Environmental Chemistry: Pollution and Remedial Perspective</i>, Narosa Publication, 2017.</p> <p>7. A.K. De, <i>Environmental Chemistry</i>, New Age, 2006.</p> <p>8. A.C. Stern, R.W. Boubel, <i>Fundamentals of Air Pollution</i>, D. Bruce turner &amp; D.L.Fox, Academic Press, 1984.</p> <p>9. R.A. Horne, <i>Chemistry of Our Environment</i>”, John Wiley, N.Y. (1978).</p> <p>10. C.N. Sawyer &amp; P.J. Macarty, <i>Chemistry for Environmental Engineering</i>, Mc Graw Hill, 1978.</p> <p>12. L.L. Ciaccio, <i>Water and Water Pollution Hand Book</i>”, Marcel Dekker, 1973.</p> <p>13. J.C. Lamb, <i>Water Quality and its Control</i>, John Wiley &amp; Sons, N.Y., 1985.</p>	